

GOLD ASSAY (FIRE ASSAY METHOD)

This Standard prescribed the cupellation or fire assay method for assaying of gold jewellery/ artefacts covered in IS 1417.

GOLD ASSAY (FIRE ASSAY METHOD)

Terminology:

Fine Gold : -It is gold having fineness(part per 1000 ,represented as ‰) 999 and above without any negative tolerance and free from Pb, Hg, Bi, Se and platinum group metals.

Cupellation:-An oxidizing fusion of lead, gold ,silver and associated base metal in a cupel which absorbs the lead oxide along with base metals oxides leaving a lead of gold and silver (along with platinum group metals, if any) on the cupel.

Parting:-Separating of silver from gold by selectively dissolving the silver in acid , using nitric acid.

GOLD ASSAY (FIRE ASSAY METHOD)

- STEP 1: Melting Process

Put the scrapped metal part of jewellery in charcoal crucible and covered with charcoal powder.

- Put the crucible in melting furnace for 3-5 minutes at the temperature of 1068-1070°C.
- Cool down the metal and make a strip using rolling mill.
- This process make alloy homogeneous in purity.



Gold Strip



Graphite Crucible

GOLD ASSAY (FIRE ASSAY METHOD)

- STEP 2:weighing
- XRF the alloy(strip) and note down the percentage of metals present in the strip.
- **Duplicate/Analysis Samples:-**
- Transfer two samples of the alloy(strip) for each lots between 125 mg to 250 mg weight to the nearest ± 0.01 mg.
- Eg 180 mg of two sample for a lot.

GOLD ASSAY(FIRE ASSAY METHOD)

- Pure gold /check gold /fine gold(999.9‰):
- Calculate weight of check gold according to the sample weight and percentage of gold present in the alloy(strip).

• Eg. If the percentage of gold in strip is 916.3(obtained in XRF)

• The check gold(in mg)= $\frac{\text{sample weight} \times \text{Purity of gold obtained in XRF}}{100}$

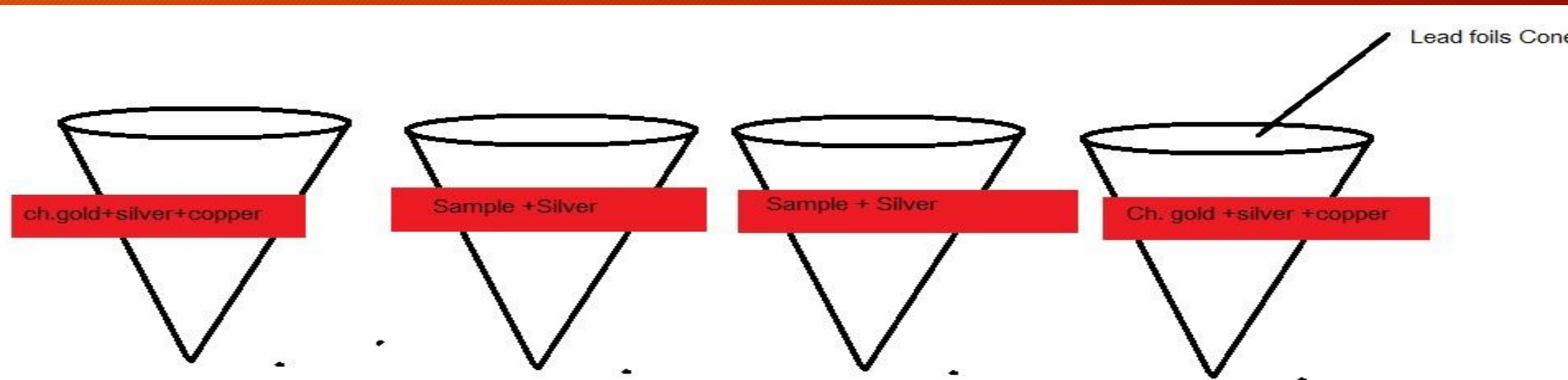
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$$= \frac{180 \times 91.7}{100}$$
$$= 165.06$$

GOLD ASSAY (FIRE ASSAY METHOD)

- **Copper** :similarly find the weight of copper using sample weight and percentage of copper present in strip.
- Eg :if there is 6% of copper in XRF
- Copper (in mg)= $\frac{180 \times 6}{100} = 7.2$
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- **Silver**: add 2.3 to 3.0 times wt. of pure gold/check gold.
- Silver (in mg)= $2.5 \times 165.06 = 412.65$.

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- **Lead** :4 gram (if the sample wt. is less than 200 mg)
- 6 gram(if the sample wt. is more than 200 mg up to 250 mg)
- Here the sample wt. is 180 mg so take 4 gm. of lead.
- Now make the open cone of the lead foil and the metal according to the figure.



GOLD ASSAY(FIRE ASSAY METHOD)

- Table for 4 sample and 2 check gold.....

sample	Au	Ag	Cu	Pb
P1(ch. gold)	165.747mg	414.35mg	7.2mg	4gm
S1 (sample)	180.564mg	414.75mg	-	4gm
S2 (sample)	180.385mg	414.82mg	-	4gm
S3 (sample)	180.653mg	414.35mg	-	4mg
S4 (sample)	180.455mg	414.33mg	-	4gm
P2 (ch. gold)	165.726mg	414.12mg	7.2gm	4gm

GOLD ASSAY (FIRE ASSAY METHOD)

- Now make the ball of each cone putting metals inside as describe in the above figure, using balling pliers .



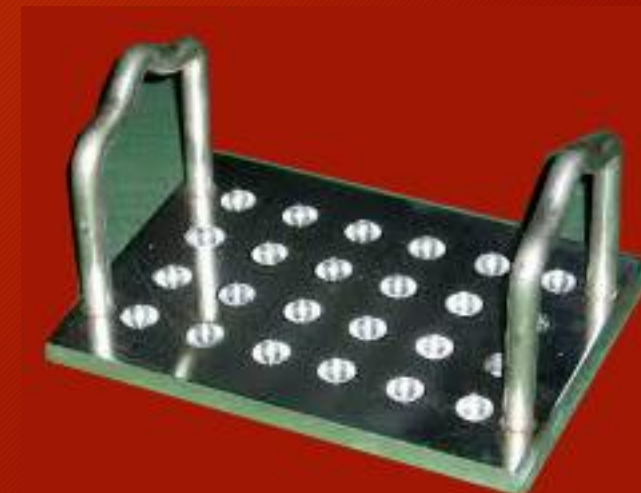
GOLD ASSAY (FIRE ASSAY METHOD)

- STEP 3: Cupellation
- Put all the balls in the pre-heated magnesia cupel in the furnace at 1100°C,
- Put this cupel inside the cupellation furnace.
- Temperature - 1050°C-1100°C
- Time : 20-30 minutes (Depending on the amount of lead used)
- Maintain oxidizing atmosphere



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- Fixed the temperature of furnace at 800°C.
- Withdrawal the cuples while temperature come down approx 800°C.
- After the oxidation is complete cool down the gold and silver button forms and clean it with brush.
- Flattered the button with hammer.
- Anneal the flattered button to red heat at 700°C for 5 minutes using annealing tray on annealing furnace



GOLD ASSAY (FIRE ASSAY METHOD)

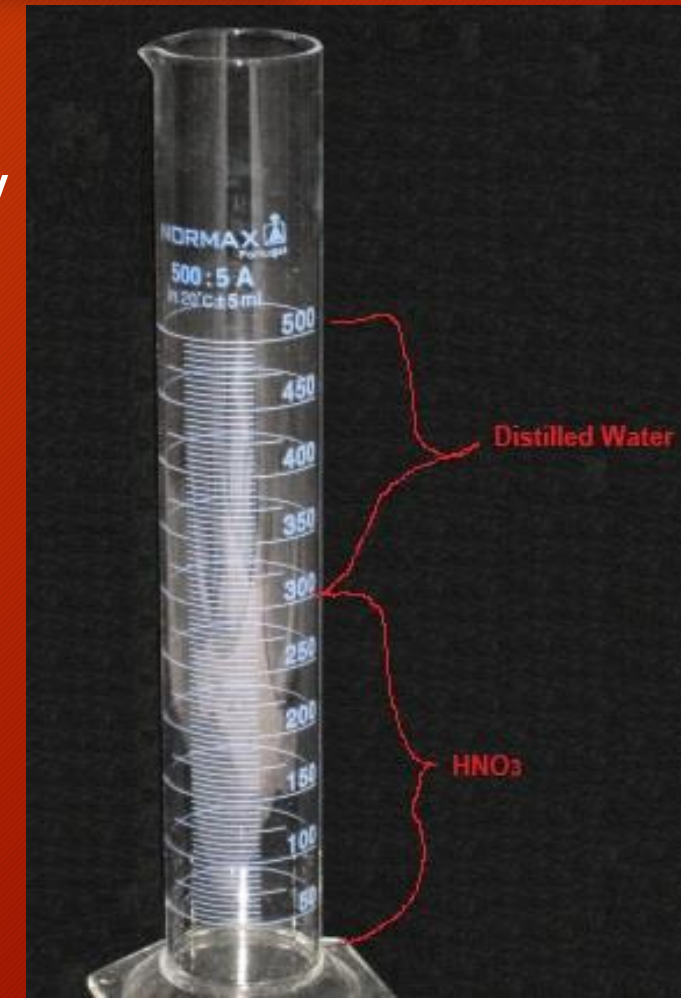
- Roll them to form strip of 0.12 to 0.15 mm on rolling mill
- Anneal the rolled strip.
- At 700°C for 5 minutes.
- Roll them to form cornet.
- Keep the rough surface outside without touching the surface of each other layer.



GOLD ASSAY (FIRE ASSAY METHOD)

- Parting process:
- Parting 1: Prepare Dilute nitric acid of 1.2 specific gravity (take 200ml of nitric acid+300ml distilled water).

Parting 2: Prepare Dilute nitric acid of 1.3 specific gravity (take 300ml of nitric acid +200ml distilled water).



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- Now heat the hot plate at 92-95°C (below 100°C)
- Dip the parting tray with cornet in parting flask containing parting:1 chemical for 15 minutes.
- Silver dissolve in nitric acid .
- $3\text{Ag}(s) + 4\text{HNO}_3(\text{aq}) \rightarrow 3\text{AgNO}_3(\text{l}) + \text{NO}(\text{g}) + 2\text{H}_2\text{O}$
- Now wash the tray with hot distilled water.
- Again repeat the same process in parting :2 for 10 minutes.
- Wash 2 times with distilled water.
- Annealing of cornet: 700°C for 5 minutes.



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- Weight the cornet and note according to their sample number
- Calculations:

Sample No./Code no.	Initial wt. of sample(mg) m1	Wt. of gold cornet after assaying(mg) m2	Δ (mg) M1-m2=	Avg Δ (mg)
C1 (check gold)	165.747	165.763	$\Delta_1 = -0.012$	
S1	180.564	165.523 ← -S1'		$(\Delta_1 + \Delta_2) / 2$
S2	180.385	165.340 ← -S2'		
s3	180.563	165.485 ← -S3'		$= -0.010$
S4	180.455	165.368 ← -S4'		
C2 (check gold)	165.726	165.806	$\Delta_2 = -0.008$	

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- Fineness of Sample:

- sample ppt =
$$\frac{\text{Final Cornet Wt.} + \text{avg } \Delta}{\text{Initial sample wt.}} \times 1000$$

- S1 ppt = $[\{165.523 + (-0.010)\} / 180.564] \times 1000$

- = 916.64 \approx 916.6

- Similarly

- S2 ppt = 916.5,

- S3 ppt = 916.4,

- S4 ppt = 916.3.

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- Result:

Sample of first lot are -
S1 ppt=916.6
S2 ppt=916.5
Sample of second lot are -
S3 ppt=916.4
S4 ppt=916.3



- If the difference in fineness is less than 0.5 → result OK
- If the difference in fineness is greater than 0.5 → Repeat the assay
- Now average fineness for first lot is $(916.6+916.5)/2=916.55\approx 916.5$
- Now average fineness for second lot is $(916.4+916.3)/2=916.35\approx 916.3$