

Lecture 3: Exercise on measurement of quantities.

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Exercise – (I) specific gravity of slurry

Calculate specific gravity of slurry which contains 65% lime. The specific gravity of lime is 2600 kg/m³. Also calculate volume of slurry.

Substituting the values in equation 1 of lecture 2

$$65 = \frac{100 \times 2600 (\rho_m - 1000)}{\rho_m (2600 - 1000)}$$

By solving $\rho_m = \text{specific gravity of slurry} = 1667 \frac{\text{kg}}{\text{m}^3}$

$$\text{Volume of slurry in percent} = 65 \times \frac{1667}{2600} = 41.66\%$$

Exercise – (II): % solids in slurry

A slurry stream containing quartz is diverted into a 1 liter can. Time taken to fill is 8 seconds. The density of the slurry and quartz is 1400 kg/m³ and 2600 kg/m³ respectively. Calculate (a) percent solids by weight and (b) mass flow rate quartz within slurry.

Substituting the values in equation 1 of lecture 2

$$\begin{aligned} \text{\% solid (\%x)} &= \frac{100 \times \rho_s (\rho_m - 1000)}{\rho_m (\rho_s - 1000)} \\ &= \frac{100 \times 2650 \times 400}{1400 \times 1600} \\ &= 45.88\%. \end{aligned}$$

Equation 3 of lecture 2

$$M(\text{kg/hr}) = \frac{F \rho_s (\rho_m - 1000)}{(\rho_s - 1000)}$$

$$F = \frac{3600}{8000} \text{ m}^3/\text{hr}.$$

$$M = \frac{3600}{8000} \times 1400 \times 0.4588$$

$$=289\text{kg/hr}.$$

If the time to fill the can is 7s, rest everything remains the same, calculate mass flow rate of quartz in slurry.

Answer =330 kg/hr

Exercise (III) Mixing of slurry streams

Two slurry streams enter a pump. One stream has flow rate of $5\text{m}^3/\text{hr}$ and contains 40% solids by weight. Other stream has $3.4\text{ m}^3/\text{hr}$ flow rate and contains 55% solids by weight. Density of solid is 3000kg m^{-3} in both slurry streams. Calculate tonnage of any solids pumped/hr.

To solve the problem, use the following steps.

1. Calculate density of slurry streams $\rho_{M1} = 1364\text{ kg/m}^3$ and $1579\frac{\text{kg}}{\text{m}^3}$
2. Calculate mass flow rate of slurry $M_1 = 2728$ and $M_2 = 2953\text{ kg/hr}$
3. Add mass flow rate which will give 5.681T/hr .

Exercise (IV) Slurry making

Calculate how many kg of magnetite must be added to 100kg water to make up a slurry with specific gravity (ρ_m) = 1.4 g/cm^3 .

Specific gravity of $\text{Fe}_3\text{O}_4 = 5.2\text{g/cm}^3$

$$\text{Wt \% solid} = \frac{100 \times 5.2 (1.4 - 1)}{1.4 \times (5.2 - 1)} = \frac{100 \times 5.2 \times 0.4}{1.4 \times 4.2} = 35.4\%$$

$$\text{Wt \% water} = 100 - 35.4 = 64.6\%$$

$$\frac{\text{total mass}}{\text{mass of water}} = \frac{M_m}{M_{\text{H}_2\text{O}}} = \frac{100\%}{64.6\%}$$

$$M_m = \frac{100\%}{64.6\%} \times 100$$

$$=155\text{kg}$$

$$\text{Mass of magnetite} = 155 - 100 = 55\text{kg}$$

Exercise (V) Units

- a) Convert 360 mm Hg pressure into (a)N/m² and (b)Lb/ft²

Answer = 47995.2 N/m² and 1002.4 Lb/ft²

- b) Calculate the value of universal gas constant into (a)CGS unit and (b)MKS unit . given

$R = 82.0578 \frac{\text{Cm}^3 \text{atim}}{\text{gmole } ^\circ\text{C}}$. The problem is solved in lecture 2.

- c) Pressure in an evacuated vessel was recorded 1.2dynes/Cm². Express this value in microns of mercury at 25°C.

Answer 1.596 micron.

Exercise (VI) Quiz question

Two streams of slurry enter a pump. The volumetric flow rate of stream2 is 0.68 times and density is 1.16 times that of slurry no. 1 respectively. if density of slurry no.1is 1364 kg/m³ . Then mass flow rate (M₁) of slurry stream no.1is

- (A) M₁ = 0.72 M₂ (B) M₁ = 0.92 M₂ (C) M₁ = 1.5 M₂
(D) M₁ = 0.8 M₂

M₂ is mass flow rate of slurry 2

One can derive the following equation expressing mass flow rates in terms volume flow rates and density of slurry

$$\frac{M_1}{M_2} = \frac{F_1}{F_2} \frac{(\rho_{M1}-1000)}{(\rho_{M2})}$$

M₁ and M₂ mass flow rate of stream 1 and 2

F₁ and F₂ volume flow rate of stream 1 and 2

ρ_{M1} and ρ_{M2} density of slummy 1 and 2.

Using above equation: F₂ = 0.68 F₁ and ρ_{M2} = 1.16 ρ_{M1} .

M₁ = 0.92 M₂ 'B' is correct.